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S Supporting Information

[ABSTRACT:](#page-9-0) For the efficient approach to medicinally important α-branched 3-acyltetramic acids, the key reaction of O- to C- acyl rearrangement using α -amino-acid-derived 4-O-acyltetramic acids was extensively examined in the presence of various metal salts. Use of $CaCl₂$ or NaI dramatically changed the results in the reaction efficiency and rapidly brought about the desired α -branched 3-acyltetramic acids in markedly improved yields. We also discuss an epimerization at C5 stereocenter under the rearrangement conditions as well as the tolerance for structural variation at C3 and C5. In addition to the preceding success in the total synthesis of

new cytotoxic tetramic acid, penicillenol A_1 , this methodology could be also applied to the first total synthesis of penicillenol A_2 .

■ INTRODUCTION

2,4-Pyrrolidinedione derivatives, so-called tetramic acids, can serve as one of the basic building blocks for bioactive molecules found in a variety of natural products from marine organisms, fungi, and bacteria. 1 Among these compounds, 3-acylsubstituted tetramic acids (Figure 1) have proven particularly appealing to the medi[ci](#page-9-0)nal community because of their remarkable potency such as antiviral and an[ti](#page-1-0)tumor activities (tenuazonic acid),^{2a} herbicidal activity (macrocidin A),^{2b} inhibitory activities against bacterial RNA polymerases (streptolydigin),^{2c,d} anti-HIV activit[y](#page-9-0) (Sch 213766),^{2e} and cytotoxicity (p[enic](#page-9-0)illenols $A_{1,2}$).^{2f} In particular, extensive research focused on streptoly[digi](#page-9-0)n, whose outstanding inhibitor[y a](#page-9-0)ctivity against bacterial RNA poly[m](#page-9-0)erase has received particular attention over the past decade, provided an important stimulus for understanding mechanistic complexities of the biological events.³ Despite the great significance of 3-acyltetramic acids, development of comprehensive and efficient protocols for this [t](#page-9-0)ype of compound is substantially less advanced, because there is an intrinsic difficulty in accessing this structural system.^{4−10} In this context, one of the conventional strategies for the preparation of this class of compounds has utilized Dieckman[n-](#page-9-0)t[yp](#page-9-0)e cyclization of β -ketoamides A (Figure 2).^{4,5} This synthetic method faces the risk of epimerization at the C5 stereocenter, which would be attributed to enolization [eq](#page-1-0)[uili](#page-9-0)brium under the basic conditions, thus making this methodology unsuitable for stereoselective synthesis.^{5d,e,h,j,l} Another possible synthetic strategy involves direct acylation of C3-unsubstituted tetramic acids B with the use of a[cid chl](#page-9-0)orides.⁶ This has a limited functional group tolerance, because harsh conditions of operation employing Lewis acids such as BF_3 ·OEt₂ are essential to drive the reaction smoothly. Alternatively, a synthetic approach based on O- to

C-acyl rearrangement of 4-O-acyltetramic acids C has been demonstrated to enable indirect installation of acyl groups into the tetramic acid ring systems.⁷ This methodology, however, has limited applicability only for the reactions of the substrates bearing linear migratory segme[n](#page-9-0)ts that would be preferred on steric grounds^{6f,7d} and would be incompatible with complicated migratory segments such as α -branched alkylcarbonyl groups. In this regard[, we](#page-9-0) have recently overcome this problem by the use of $CaCl₂$ as an effective additive, giving a marked improvement in the reaction efficiency for the formation of the 3 acyltetramic acid bearing α -methyloctanoyl moiety. Moreover, we accomplished the first total synthesis of penicillenol A_1 through N-methylation of amide and subsequent removal of TBS protecting group (Scheme 1).⁸ Despite less accumulated information available for the synthesis of 3-acyltetramic acid derivatives, our previous work g[av](#page-1-0)[e](#page-9-0) some indication of a rich opportunity for accessing diverse 3-acyltetramic acids with the other branched segments. In an attempt to examine the validity and generality of this synthetic methodology and to expand its utility, we explored the scope of the reaction with a diverse set of the substrates. In the present article, we provide a full account of our efforts in developing the synthetic approach to variously substituted 3-acyltetramic acids and demonstrate the versatility for the remaining synthetic challenge associated with the total synthesis of naturally occurring antibacterial penicillenol A_2 .

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Figure 1. Structures of naturally occurring 3-acyltetramic acids.

Figure 2. Conventional synthetic approaches to 3-acyltetramic acids.

RESULTS AND DISCUSSION

4-O-Acyltetramic acids with a variety of substituents 5a−n were prepared as shown in Figure 3. $^{65,7{\rm d},11}$ A series of 3-unsubstituted tetramic acids 4a−e obtained through condensation−decarboxylation sequences in good yiel[d](#page-2-0)[s from](#page-9-0) N-Boc or N-Cbz protected amino acids 3a−e ¹² could be condensed with a range of carboxylic acids in the presence of DCC and DMAP to provide 5a−m. For the preparation [of](#page-9-0) $5n$, esterification of $4b$ using Piv₂O as a condensing agent was preferred over the above synthetic protocol, affording the best yield of the desired product.

With these compounds in hand, we turned our focus toward investigating the O- to C-acyl rearrangement of 5 to obtain 3 acyltetramic acids 6. To optimize this process, a similar series of experiments were undertaken with 5a under various reaction conditions (Table 1). Without any metal salts, the reaction at room temperature led to give only a 10% yield of the desired product 6a for a [pe](#page-2-0)riod of 24 h along with recovery of the starting material (entry 1). In contrast, the reaction in the presence of LiBr (1.5 equiv) showed complete consumption of the starting material and gave distinctly better yield of 6a (43%, entry 2).

Similarly, such effects could be observed for the reactions with the use of the other metal salts such as NaCl (entry 3), KF (entry 6), or $CeCl₃$ (entry 8) to afford 6a in moderate yields, while $FeCl₃$ would be detrimental to the transformation to result in decomposition of the substrate (entry 9). Among the various metal salts examined, $CaCl₂$ (71%, entry 7) and NaI (70%, entry 4) exhibited the marked performances to generate 6a, and we therefore conclude that these materials would be the optimal additives to promote the O- to C-acyl rearrangement processes efficiently. In an analogy to closely related systems, the mechanism of rearrangement could be understood on the basis of primary process, namely, a C−O bond-cleavage and recombination of the migrating ion pairs.¹³ The presence of ionic species in the metal salts may help in stabilization of the in situ generated ion pairs that would be bou[nd](#page-9-0) together by weak Coulombic attraction, thus facilitating the formation of 3 acyltetramic acids.

As seen in Figure 4, a wide variety of 4-O-acyl tetramic acids 5 were shown to undergo the intramolecular transformations in the presence of $CaCl₂$ to give the corresponding 3-acyltetramic acids 6. Indeed, the substrates bearing the substituted benzoyl and α -branched alkylcarbonyl moieties 5a–m showed remarkably high reactivities to generate 6a−m in satisfactory yields (51−79%), respectively, whereas the sterically demanding tertbutyl analogue 5n failed to afford the desired compound, decomposing to 4n under conditions identical to those for the other substrates. Furthermore, no obvious difference was observed in the reaction efficiency when the C5-substituent on the tetramate core was varied, and all the series of 4-O-acyl tetramic acids derived from isoleucine 5b−g, phenylalanine 5h, alanine 5i, serine 5j, and threonine 5k−m gave the corresponding rearrangement products 6b−m in moderate to good yields. These results suggest that the course of the reaction is not influenced by the structural properties of the tetramate cores but rather by those of the acyl groups on the

Figure 3. Preparation of 4-O-acyltetramic acids 5. Isolated yields of products are indicated below each entry. *See ref 11.

Table 1. O- to C-Acyl Rearrangement of 5a in the Presence of Various Metal Salts

oxygen atoms. As for stereochemical integrity of the reaction products, it should be noted that small amounts of the epimeric C5 diastereomers were in situ genera[ted](#page-9-0) for the cases of the substituted benzoyl series 6b–d, as revealed by the ¹H NMR analysis. Apparently, the occurrence of the epimerization can be understood in terms of pathways involving deprotonation of the C5−H and enolization to form planar structures of the relevant enolates. Through inspection of the stereochemical control for the product formation, a higher degree of epimerization was observed as an increase in the reaction time. Actually, prolongation of the reaction time to ensure complete consumption of the reactants led to increases in the relative ratios of the epimeric isomers over the desired ones, where 5c representing the typical example gave a ratio of 70:30 (6c/epimer) for 16 h as compared with 78:22 for 1 h along with $5b$ $(83:17$ for 3 h) and $5d$ $(87:13$ for 3 h). On the other hand, the series of 4-O-alkylcarbonyl tetramic acids 5e−m underwent fast reactions that completed within 1 h, affording the enantiomerically pure products. In such cases, the epimerization was exclusively avoided, probably because of substantially much slower rates as compared to the O- to C-acyl rearrangement processes. In this regard, we found that a noticeable epimerization of the recovered substrate took place

Figure 4. Acyl rearrangement of 4-O-acyltetramic acids in the presence of CaCl₂. Isolated yields are indicated below each entry. *Reaction time and diastereomeric ratio determined by ¹H NMR are indicated in parentheses.

at a rate of 96:4 (5b/epimer) when the enantiomerically pure 5b was treated with $Et₃N$ (1.5 equiv) and in the presence of $CaCl₂$ (1.5 equiv) in $CH₂Cl₂$ at room temperature,¹⁴ as observed by ¹H NMR, while no epimerization of the product occurred when 6b was treated under the same cond[iti](#page-9-0)ons (Figure 5). These findings confirm that the formation of

Figure 5. Comparison of ${}^{1}\mathrm{H}$ NMR spectra on H5 of 5b or 6b (300 MHz, CDCl₃, δ in ppm). (a) **5b**, (b) **5b** after treatment with Et₃N (1.5) equiv) and CaCl₂ (1.5 equiv) in CH₂Cl₂ at room temperature, (c) 6b, and (d) 6b after treatment with Et_3N (1.5 equiv) and $CaCl₂$ (1.5 equiv) in $CH₂Cl₂$ at room temperature.

3-acyltetramic acids competes with the epimerization proceeding at comparable rates. From the above considerations, we can conclude that the metal salts serve a dual purpose of suppressing the epimerization under the conditions employed as well as enhancing the product formation.¹⁵

As our continuing interest in the context of the present studies, we considered that a properly desi[gne](#page-9-0)d 3-acyltetramic acid obtained through the O- to C-acyl rearrangement would serve as a substrate pertinent to achieving the total synthesis of natural products such as penicillenols, which have been isolated from the culture broth of a fungus Penicillium sp. as cytotoxic agents. $2f$ In our previous work, the total synthesis of penicillenol A_1 1 and its C9-epimer has been completed by emplo[yin](#page-9-0)g 6l,m, respectively.⁸ Following the above synthetic strategy, we considered that the synthesis of penicillenol A_2 , a C5-epimer of penicillenol A_1 , would be achieved using D-allothreonine as a starting chiral source. Thus, we turned to synthesize this natural product to demonstrate the versatility of this synthetic methodology for a generous supply of 3-acyltetramic acids (Scheme 2). The synthesis started with a 56:44 mixture of L-threonine and D-allothreonine, which could be prepared by isomerizatio[n](#page-4-0) of L-threonine.¹⁶ When this material was allowed to react under conditions identical to that for 4e, a 61:39 inseparable diastereomeric mi[xtu](#page-9-0)re of 9 was obtained. Subsequent condensation of this material with (S)-2 methyloctanoic acid^{17} afforded a mixture consisting of 10 and its diastereoisomeric counterpart 5m, which could be separated by silica gel colu[mn](#page-9-0) chromatography. The enantiomerically

pure 10 underwent the O- to C-acyl rearrangement followed by N-methylation to produce 12 in 58% yield over two steps. At the final step, the TBS group of 12 could be removed by exposure to methanolic HCl (2%) to give 2 in 65% yield as a 90:10 mixture of Z:E isomers.

Identification of this product was performed by spectroscopic analyses and optical rotatory measurements. As a consequence, 1 H NMR, 13 C NMR, and IR spectroscopic data of 2 appeared to be fully consistent with those of the natural penicillenol A_{2} , although the optical rotation of 2 $\{[\alpha]_{\mathbb{D}}^{22} +49.2$ (c 0.137, MeOH)} showed a significant difference to that reported in the literature^{2f} $\{ [\alpha]_{\mathbf{D}}^{25} + 386.7 \ (c \ 0.135, \ \text{MeOH}) \}.$ ¹⁸ Unfortunately, other data to be discussed were unavailable except that the optical r[ot](#page-9-0)ations of both penicillenols fluctu[ate](#page-9-0)d significantly because of their low solubilities in MeOH.¹⁹ Further measurements of the synthetic sample of 1 and 2 in CHCl₃ allowed us to reevaluate the optical activities, giving [c](#page-10-0)onstant values of $[\alpha]_{\text{D}}^{23}$ –47.6 (c 0.918) for 1 and $[\alpha]_{\text{D}}^{22}$ +27.8 (c 0.864) for 2. Although an obvious reason for the observed difference in the $[\alpha]_{\text{D}}$ values remains unclear at present, we are reasonably confident with the complete spectroscopic characterization that the synthetic approach described above should ensure completion of the first total synthesis of penicillenol A_2 with the complete structure and absolute stereochemistry.²⁰

■ CONCLUSION

The synthesis of 3-acyltetramic acids via O- to C-acyl rearrangement has been systematically investigated. We demonstrated that the presence of metal salts significantly improved the product yields to produce the desired 3-acyltetramic acids, and a variety of substituents were found to be well tolerated. Coupled with our preceding report on the success in synthesizing penicillenol A_1 , this synthetic method has been demonstrated to allow for completing the first total synthesis of penicillenol A_2 .

EXPERIMENTAL SECTION

(S)-5-((S)-sec-Butyl)-4-oxo-2-pyrrolidone (4b). To a solution of Meldrum's acid (1.20 g, 8.33 mmol) and protected amino acid 3c (1.76 g, 7.61 mmol) in dichloromethane (50 mL) at 0 °C were added 1-ethyl-3-(3-dimethylaminopropyl)carbodiimide hydrochloride (EDCI) (1.75 g, 9.13 mmol) and DMAP (1.39 g, 11.4 mmol). After stirring for 5 h at room temperature, the reaction mixture was poured into EtOAc (150 mL) and extracted. The organic layer was washed with brine and 5% aqueous $NaHSO_4$ solution, dried over $Na₂SO₄$, and filtered. The filtrate was refluxed for 1 h until completion of the reaction (monitored by TLC) and concentrated under reduced pressure. The residue was treated with 1:1 TFA/dichloromethane (2.0 mL) for 1 h. After removal of the solvent by evaporation, the crude product

was purified by column chromatography (silica gel, hexane/EtOAc = 1/1) to give 4b (1.01 g, 86%) as a white powder: dp 94−100 °C; $[\alpha]_D^{27}$ –71.7 (c 1.04, CHCl₃); IR (KBr) 3178 (N–H), 3098 (N–H), 2936 (C−H), 1772 (C=O), 1694 (C=O) cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 6.59 (brs, 1H), 3.92 (d, J = 4.2 Hz, 1H), 2.97 (s, 2H), 1.89 (m, 1H), 1.46−1.26 (m, 2H), 1.00 (d, J = 6.9 Hz, 3H), 0.92 (t, J = 7.5 Hz, 3H); 13C NMR (75 MHz, CDCl3) δ 207.8 (C), 172.5 (C), 69.1 (CH), 41.6 (CH), 37.6 (CH), 24.2 (CH₂), 15.1 (CH₃), 11.4 (CH₃). Anal. Calcd for $C_8H_{13}NO_2$: C, 61.91; H, 8.44; N, 9.03. Found: C, 62.02; H, 8.58; N, 8.64.

(S)-5-((tert-Butyldimethylsilyloxy)methyl)-4-oxo-2-pyrrolidone (4d). To a solution of Meldrum's acid (951 mg, 6.60 mmol) and protected amino acid 3d (2.12 g, 6.00 mmol) in dichloromethane (30 mL) at 0 °C were added EDCI (1.38 g, 7.20 mmol) and DMAP (1.10 g, 9.00 mmol). After stirring for 2 h at room temperature, the reaction mixture was filtered through a pad of Celite. The filtrate was washed with 5% aqueous $NaHSO₄$ solution, water, and brine. The organic phase was dried over $Na₂SO₄$, filtered, and evaporated to give the crude mixture. This material was dissolved in EtOAc (90 mL), and the solution was refluxed for 1 h until completion of the reaction (monitored by TLC) and concentrated under reduced pressure. A suspension containing the residue and Pd/C (31.2 mg) in EtOH (9.0 mL) was stirred under hydrogen atmosphere for 2 h. The reaction mixture was filtered through a pad of Celite, and the residue was washed with EtOAc. The filtrate was concentrated under reduced pressure, and the residue was purified by column chromatography (silica gel, CHCl₃/MeOH = $40:1$) to give 4d (891 mg, 61%) as a pale yellow solid: dp 78−84 °C; $\left[\alpha\right]_D^{\text{ }27}$ −53.7 (c 1.03, CHCl₃); IR (KBr) 3200 (N−H), 3104 (N−H), 2930 (C−H), 1778 (C=O), 1690 (C= O) cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 7.43 (brs, 1H), 4.01 (brt, J = 3.3 Hz, 1H), 3.92 (dd, J = 3.3, 10.5 Hz, 1H), 3.77 (dd, J = 2.4, 10.5 Hz, 1H), 2.91 (s, 2H), 0.84 (s, 9H), 0.03 (s, 3H), 0.02 (s, 3H); 13C NMR (75 MHz, CDCl₃) δ 206.5 (C), 172.3 (C), 66.0 (CH₂), 63.0 (CH), 41.4 (CH₂), 25.6 (CH₃), 18.0 (CH₃), -5.8 (CH₃), -5.9 (CH₃). Anal. Calcd for $C_{11}H_{21}NO_3Si$: C, 54.29; H, 8.70; N, 5.76. Found: C, 54.44; H, 8.43; N, 5.93.

(S)-5-((R)-1-(tert-Butyldimethylsilyloxy)ethyl)-4-oxo-2-pyrrolidone (4e). To a solution of Meldrum's acid (1.50 g, 10.3 mmol) and protected amino acid 3e (3.45 g, 9.39 mmol) in dichloromethane (35 mL) at 0 °C were added DCC (2.13 g, 10.3 mmol) and DMAP (1.49 g, 12.2 mmol). After stirring for 10 h at 0 $^{\circ}$ C, the reaction mixture was filtered through a pad of Celite, and the residue was washed with CH_2Cl_2 . The filtrate was washed with 5% aqueous NaHSO₄ solution and brine. The organic phase was dried over $Na₂SO₄$, filtered, and evaporated to give the crude mixture. The solution of this material in EtOAc (200 mL) was refluxed for 5 h until completion of the reaction (monitored by TLC). After removal of the solvent by evaporation, the suspension containing the residue and Pd/C (500 mg) in EtOAc (50 mL) was stirred under hydrogen atmosphere for 12 h. The mixture was filtered through a pad of Celite, and the residue was washed with EtOAc. The filtrate was concentrated under reduced pressure, and the residue was purified by column chromatography (silica gel, hexane/ EtOAc = $3/1-2/1-1/1$) to give 4e (911 mg, 38%, 3 steps) as a white

powder: dp 122−124 °C; $[\alpha]_D^2$ ⁹ −84.8 (c 1.10, CHCl₃); IR (NaCl) 3199 (N−H), 3111 (N−H), 1777 (C=O), 1709 (C=O) cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 7.90 (brs, 1H), 4.21 (dq, J = 2.7, 6.6 Hz, 1H), 3.80 (brd, J = 2.6 Hz, 1H), 2.90 (s, 2H), 1.24 (d, J = 6.6 Hz, 3H), 0.82 (s, 9H), 0.04 (s, 3H), −0.02 (s, 3H); 13C NMR (75 MHz, CDCl3) δ 206.8 (C), 172.9 (C), 70.2 (CH), 68.1 (CH), 41.3 (CH₂), 25.5 (CH₃), 20.3 (CH₃), 17.7 (C), −4.5 (CH₃), −5.4 (CH₃). Anal. Calcd for C₁₂H₂₃NO₃Si: C, 55.99; H, 9.01; N, 5.44. Found: C, 55.66; H, 8.75; N, 5.59.

General Procedure for the Preparation of the 4-O-Acyltetramic Acids. To a solution of carboxylic acid (1.1−1.4 equiv) in dichloromethane at 0 °C were added DCC (1.2−1.9 equiv) and DMAP (0.2−0.3 equiv). After stirring for 5 min at the same temperature, 4 (1.0 equiv) was added. It was stirred at room temperature until completion of the reaction (monitored by TLC). The reaction mixture was filtered through a pad of Celite, and the residue was washed with EtOAc. The filtrate was washed with saturated aqueous NH₄Cl solution and brine. The organic phase was dried over $Na₂SO₄$, filtered, and concentrated under reduced pressure. Purification of the crude product afforded 4-O-acyltetramic acid 5.

(S)-2-Benzyl-5-oxo-2,5-dihydro-1H-3-pyrrolyl Benzoate (5a). The crude product was rinsed with $Et₂O$ (5 mL) to give 5a (40.1 mg, 86%) as a white powder: dp 113−115 °C; $[\alpha]_D^{30}$ −34.3 (c 0.978, CHCl3); IR (KBr) 3193 (N−H), 3075 (N−H), 2895 (C−H), 1752 (C=O), 1696 (C=O), 1612 (C=C) cm⁻¹; ¹H NMR (300 MHz, CDCl3) δ 8.00 (m, 2H), 7.67 (m, 1H), 7.51 (m, 2H), 7.32−7.22 (m, 5H), 6.55 (s, 1H), 6.21 (s, 1H), 4.59 (dd, J = 5.1, 8.1 Hz, 1H), 3.22 $(dd, J = 5.1, 13.5 Hz, 1H), 2.91 (dd, J = 8.1, 13.5 Hz, 1H);$ ¹³C NMR (75 MHz, CDCl₃) δ 172.9 (C), 165.8 (C), 162.2 (C), 136.0 (C), 134.5 (C), 130.3 (CH), 129.2 (CH), 128.9 (2CH), 128.0 (CH), 127.2 (CH), 108.0 (CH), 100.6 (CH), 59.0 (CH), 38.7 (CH₂). Anal. Calcd for C18H15NO3: C, 73.71; H, 5.15; N, 4.78. Found: C, 74.02; H, 5.45; N, 5.17.

(S)-2-((S)-sec-Butyl)-5-oxo-2,5-dihydro-1H-3-pyrrolyl Benzoate (5b). The crude product was purified by column chromatography (silica gel, hexane/EtOAc/AcOH = $66/33/1$) to give 5b (106 mg, 91%) as a white powder: dp 106−114 °C; $\left[\alpha\right]_D$ ²⁷ +1.97 (c 1.05, CHCl3); IR (KBr) 3198 (N−H), 3072 (N−H), 2935 (C−H), 1765 (C=O), 1682 (C=O), 1616 (C=C) cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 8.10 (m, 2H), 7.93 (brs, 1H), 7.68 (t, J = 7.5 Hz, 1H), 7.53 $(t, J = 7.5$ Hz, 2H), 6.28 (s, 1H), 4.38 (brd, $J = 3.0$ Hz, 1H), 1.97 (m, 1H), 1.44 (m, 1H), 1.24 (m, 1H), 1.09 (d, $J = 6.9$ Hz, 3H), 0.93 (t, $J =$ 7.4 Hz, 3H); 13C NMR (75 MHz, CDCl3) δ 174.4 (C), 165.5 (C), 162.3 (C), 134.4 (C), 130.2 (CH), 128.9 (CH), 128.2 (CH), 108.7 (CH), 63.1 (CH), 36.3 (CH), 23.1 (CH₂), 15.8 (CH₃), 11.9 (CH₃). Anal. Calcd for $C_{15}H_{17}NO_3$: C, 69.48; H, 6.61; N, 5.40. Found: C, 69.84; H, 6.56; N, 5.55.

(S)-2-((S)-sec-Butyl)-5-oxo-2,5-dihydro-1H-3-pyrrolyl 4- Methylbenzoate (5c). The crude product was purified by column chromatography (silica gel, hexane/EtOAc = $3/1-1/1$) to give 5c (113 mg, 91%) as a white powder: dp 112−117 °C; $[\alpha]_D^2$ ⁷ −2.51 (c 1.05, CHCl3); IR (KBr) 3241 (N−H), 3092 (N−H), 2930 (C−H), 1751 (C=O), 1690 (C=O), 1616 (C=C) cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 7.98 (d, J = 8.3 Hz, 2H), 7.34–7.30 (m, 2H), 6.25 (s, 1H), 4.36 (brd, J = 3.3 Hz, 1H), 2.46 (s, 3H), 1.96 (m, 1H), 1.44 (m, 1H), 1.26 (m, 1H), 1.09 (d, J = 6.9 Hz, 3H), 0.92 (t, J = 7.5 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 174.1 (C), 165.6 (C), 162.3 (C), 145.6 (C), 130.3 (CH), 129.7 (CH), 125.5 (C), 108.6 (CH), 63.0 (CH), 36.4 (CH), 23.1 (CH₂), 21.7 (CH₃), 15.9 (CH₃), 11.9 (CH₃). Anal. Calcd for C₁₆H₁₉NO₃: C, 70.31; H, 7.01; N, 5.12. Found: C, 69.92; H, 7.00; N, 5.12.

(S)-2-((S)-sec-Butyl)-5-oxo-2,5-dihydro-1H-3-pyrrolyl 4- Chlorobenzoate (5d). The crude product was purified by column chromatography (silica gel, hexane/EtOAc = $1/1$) to give 5d (120 mg, 91%) as a white powder: dp 88–92 °C; $\left[\alpha\right]_D{}^{29}$ –3.6 (c 1.03, CHCl₃); IR (KBr) 3249 (N−H), 2931 (C−H), 1770 (C=O), 1696 (C=O), 1656 (C=C) cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 8.03 (m, 2H), 7.79 (brs, 1H), 7.51 (m, 2H), 6.27 (s, 1H), 4.37 (brd, J = 3.0 Hz, 1H), 1.95 (m, 1H), 1.48−1.20 (m, 2H), 1.10 (d, $J = 6.9$ Hz, 3H), 0.93 (t, $J =$ 7.2 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 174.1 (C), 165.3 (C),

161.5 (C), 141.2 (C), 131.5 (CH), 129.4 (CH), 126.6 (C), 108.9 (CH), 63.0 (CH), 36.4 (CH), 23.1 (CH₂), 15.8 (CH₃), 11.9 (CH₃). Anal. Calcd for $C_{15}H_{16}CINO_3$: C, 61.33; H, 5.49; N, 4.77. Found: C, 61.14; H, 5.76; N, 4.82.

(S)-2-((S)-sec-Butyl)-5-oxo-2,5-dihydro-1H-pyrrol-3-yl Isobutyrate (5e). The crude product was purified by column chromatography (silica gel, hexane/EtOAc = $1/1$) to give 5e (52.9 mg, 52%) as a white powder: dp 88–92 °C; $[\alpha]_D^2$ +45.9 (c 0.880, CHCl₃); IR (KBr) $3234 (N-H)$, $3103 (N-H)$, $2935 (C-H)$, 1778 (C=O), 1687 (C= O), 1652 (C=C) cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 7.36 (brs, 1H), 6.08 (s, 1H), 4.21 (t, $J = 1.5$ Hz, 1H), 2.74 (sep, $J = 6.9$ Hz, 1H), 1.85 (m, 1H), 1.39−1.13 (m, 8H), 1.03 (d, J = 6.9 Hz, 3H), 0.90 (t, J = 7.4 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 174.2 (C), 172.6 (C), 165.5 (C), 108.2 (CH), 62.8 (CH), 36.1 (CH), 34.4 (CH), 22.9 $(CH₂)$, 18.5 (CH₃), 18.4 (CH₃), 15.8 (CH₃), 11.8 (CH₃). Anal. Calcd for $C_{12}H_{19}NO_3$: C, 63.98; H, 8.50; N, 6.22. Found: C, 64.26; H, 8.83; N, 6.59.

(S)-2-((S)-sec-Butyl)-5-oxo-2,5-dihydro-1H-3-pyrrolyl 2- Methylbutanoate (5f). The crude product was purified by column chromatography (silica gel, hexane/EtOAc = $1/1$) to give 5f (48.0 mg, 70%) as a white powder: dp 95−99 °C; IR (KBr) 3233 (N−H), 3100 $(N-H)$, 2936 (C−H), 1778 (C=O), 1686 (C=O), 1652 (C=C) cm[−]¹ ; 1 H NMR (300 MHz, CDCl3) δ 7.72 (brs, 2H), 6.09−6.08 (m, 2H), 4.20 (brt, J = 1.5 Hz, 2H), 2.57 (m, 2H), 1.84 (m, 2H), 1.74 (m, 2H), 1.59 (m, 2H), 1.38−1.13 (m, 4H), 1.25 (d, J = 7.2 Hz, 3H), 1.24 $(d, J = 6.9 \text{ Hz}, 3\text{H}), 1.04 (d, J = 6.9 \text{ Hz}, 6\text{H}), 0.97 (t, J = 7.5 \text{ Hz}, 3\text{H}),$ 0.96 (t, $J = 7.4$ Hz, 3H), 0.90 (t, $J = 7.4$ Hz, 3H), 0.89 (t, $J = 7.4$ Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 174.7 (C), 172.4 (C), 172.3 (C), 165.7 (2C), 108.3 (CH), 63.2 (CH), 41.5 (CH), 41.4 (CH), 36.3 (CH), 36.2 (CH), 26.6 (CH₂), 26.4 (CH₂), 23.0 (CH₂), 22.9 (CH₂), 16.4 (CH₃), 16.1 (CH₃), 16.0 (CH₃), 12.0 (2CH₃), 11.5 (CH₃), 11.4 (CH₃). Anal. Calcd for $C_{13}H_{21}NO_3$: C, 65.25; H, 8.84; N, 5.85. Found: C, 64.92; H, 8.49; N, 5.80.

(S)-2-((S)-sec-Butyl)-5-oxo-2,5-dihydro-1H-3-pyrrolyl Cyclohexanecarboxylate (5g). The crude product was purified by column chromatography (silica gel, hexane/EtOAc = $1/1$) to give **5g** (124 mg, 90%) as a white powder: dp 112 −117 °C; $[\alpha]_D^2$ +33.1 (c 1.02, CHCl3); IR (KBr) 3235 (N−H), 3104 (N−H), 2932 (C−H), 1773 (C=O), 1692 (C=O), 1652 (C=C) cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 7.51 (brs, 1H), 6.06 (s, 1H), 4.19 (t, J = 1.5 Hz, 1H), 2.49 (m, 1H), 2.00−1.12 (m, 13H), 1.03 (d, J = 6.9 Hz, 3H), 0.89 (t, $J = 7.4$ Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 174.4 (C), 171.5 (C), 165.6 (C), 108.1 (CH), 62.9 (CH), 43.2 (CH), 36.1 (CH), 28.6 (CH_2) , 28.4 (CH₂), 25.4 (CH₂), 25.1 (CH₂), 25.0 (CH₂), 22.8 (CH₂), 15.9 (CH₃), 11.8 (CH₃). Anal. Calcd for C₁₅H₂₃NO₃: C, 67.90; H, 8.74; N, 5.28. Found: C, 67.64; H, 8.36; N, 5.58.

(S)-2-Benzyl-5-oxo-2,5-dihydro-1H-3-pyrrolyl 2-Methylbutanoate (5h). The crude product was purified by column chromatography (silica gel, hexane/EtOAc = 1/1−1/2) to give 5h (65.0 mg, 77%) as a white powder: dp 70–76 °C; $[\alpha]_D^2$ ⁹ −41.9 (c 0.515, CHCl₃); IR (KBr) 3234 (N−H), 2974 (C−H), 1773 (C=O), 1685 (C=O), 1644 (C=C) cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 7.33– 7.26 (m, 6H), 7.20−7.18 (m, 4H), 6.37 (brs, 2H), 6.02 (s, 2H), 4.40 $(dd, J = 4.2, 5.7 Hz, 2H), 3.16 (dd, J = 4.2, 13.7 Hz, 2H), 2.76 (dd, J =$ 5.7, 13.7 Hz, 2H), 2.62−2.54 (m, 2H), 1.72 (m, 2H), 1.59 (m, 2H), 1.26 (d, $J = 6.9$ Hz, 3H), 1.25 (d, $J = 6.9$ Hz, 3H), 0.99 (d, $J = 7.4$ Hz, 3H), 0.98 (t, J = 7.5 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 172.7 (C), 172.2 (C), 165.8 (C), 136.2 (C), 129.1 (CH), 128.9 (CH), 127.3 (CH), 107.6 (CH), 58.9 (2CH), 41.2 (2CH), 38.5 (CH₂), 26.4 $(2CH₂)$, 16.1 $(2CH₂)$, 14.1 $(CH₃)$, 11.3 $(CH₂)$. Anal. Calcd for C16H19NO3: C, 70.31; H, 7.01; N, 5.12. Found: C, 69.92; H, 6.61; N, 5.30.

(S)-2-Methyl-5-oxo-2,5-dihydro-1H-3-pyrrolyl 2-Methylbutanoate (5i). The crude product was purified by column chromatography (silica gel, CHCl₃/MeOH = $30/1$) to give 5i (71.5 mg, 82%) as a white powder: dp 58−65 °C; IR (KBr) 3233 (N−H), 3108 (N−H), 2937(C-H), 1776 (C=O), 1687 (C=O), 1651 (C=C) cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 7.92 (brs, 2H), 6.03 (s, 2H), 4.25 (q, J = 6.6 Hz, 2H), 2.59 (m, 2H), 1.75 (m, 2H), 1.60 (m, 2H), 1.37 (d, J = 6.6 Hz, 6H), 1.26 (d, J = 7.2 Hz, 3H), 1.24 (d, J = 7.2 Hz, 3H), 0.98 (t, J = 7.2 Hz, 6H); ¹³C NMR (75 MHz, CDCl₃) δ 173.9 (C), 172.2 (C), 172.1 (C), 167.5 (2C), 106.3 (2C), 53.7 (2CH), 41.1 (CH), 41.0 (CH), 26.3 (CH₂), 26.2 (CH₂), 17.3 (CH₃), 16.0 (CH₃), 15.9 (CH₃), 11.1 (2CH₃). Anal. Calcd for C₁₀H₁₅NO₃: C, 60.90; H, 7.67; N, 7.10. Found: C, 60.53; H, 7.54; N, 7.23.

(S)-2-((tert-Butyldimethylsilyloxy)methyl)-5-oxo-2,5-dihydro-1H-2-pyrrolyl 2-Methylbutanoate (5j). The crude product was purified by column chromatography (silica gel, hexane/EtOAc = 1/1) to give 5j (46.0 mg, 85%) as a pale yellow oil: IR (KBr) 3173 $(N-H)$, 3086 (N−H), 2927(C−H), 1778 (C=O), 1698 (C=O), 1656 (C=C) cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 6.41 (brs, 1H), 6.28 (brs, 1H), 6.08 (s, 2H), 4.25 (m, 2H), 3.93 (dd, J = 3.3, 10.1 Hz, 2H), 3.60 (dd, J = 7.5, 10.1 Hz, 2H), 2.57 (m, 2H), 1.75 (m, 2H), 1.58 $(m, 2H)$, 1.22 (d, J = 6.9 Hz, 6H), 0.94 (t, J = 7.5 Hz, 6H), 0.86 (s, 18H), 0.04 (s, 12H); ¹³C NMR (75 MHz, CDCl₃) δ 173.6 (C), 172.1 $(2C)$, 163.5 $(2C)$, 108.4 (CH) , 63.1 $(CH₂)$, 60.0 $(2CH)$, 41.1 $(2CH)$, 26.3 (2C), 25.5 (CH₃), 18.1 (CH₂), 16.1 (CH₃), 16.0 (CH₃), 11.2 (2CH₃), -5.7 (CH₃), -5.8 (CH₃). Anal. Calcd for C₁₆H₂₉NO₄Si: C, 58.68; H, 8.93; N, 4.28. Found: C, 58.51; H, 8.58; N, 4.67.

(S)-2-((R)-1-(tert-Butyldimethylsilyloxy)ethyl)-5-oxo-2,5-dihydro-1H-3-pyrrolyl 2-Methylbutanoate (5k). The crude product was purified by column chromatography (silica gel, hexane/EtOAc = 3/1) to give 5k (77.1 mg, 73%) as a pale yellow oil: IR (KBr) 3219 $(N-H)$, 3085 (N−H), 2932 (C−H), 1779 (C=O), 1697 (C=O), 1623 (C=C) cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 6.89 (brs, 2H), 6.06 (s, 2H), 4.10−4.00 (m, 4H), 2.52 (m, 2H), 1.76 (m, 2H), 1.54 $(m, 2H)$, 1.26−1.20 $(m, 12H)$, 0.94 $(t, I = 7.5$ Hz, 6H), 0.82 $(s, 18H)$, 0.03 (s, 6H), -0.01 (s, 6H); ¹³C NMR (75 MHz, CDCl₃) δ 174.1 (C), 172.2 (C), 164.1 (2C), 108.5 (CH), 67.5 (CH), 67.4 (CH), 64.0 (CH), 41.3 (CH), 41.2 (CH), 26.3 (CH₂), 26.2 (CH₂), 25.5 (CH₃), 20.6 $(2C)$, 17.8 (CH_3) , 16.1 (CH_3) , 16.0 (CH_3) , 11.3 (CH_3) , -4.4 (CH_3) , -5.2 (CH₃) -5.3 (CH₃). Anal. Calcd for C₁₇H₃₁NO₄Si: C, 59.79; H, 9.15; N, 4.10. Found: C, 59.53; H, 8.95; N, 4.38.

(S)-2-((R)-1-(tert-Butyldimethylsilyloxy)ethyl)-5-oxo-2,5-dihydro-1H-3-pyrrolyl (S)-2-Methyloctanoate (5l). The crude product was purified by column chromatography (silica gel, hexane/ EtOAc = 3/1–1/1) to give 5l (97.7 mg, 84%) as a colorless oil: $\left[\alpha\right]_D^2$ +30.6 (c 0.660, CHCl3); IR (NaCl) 3219 (N−H), 3088 (N−H), 1776 (C=O), 1701 (C=O), 1624 (C=C) cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 6.69 (brs, 1H), 6.07 (d, J = 0.9 Hz, 1H), 4.05–3.97 (m, 2H), 2.58 (m, 1H), 1.71 (m, 1H), 1.49 (m, 1H), 1.34−1.22 (m, 11H), 1.22 (d, J = 7.2 Hz, 3H), 0.87 (t, J = 6.0 Hz, 3H), 0.83 (s, 9H), 0.04 (s, 3H), 0.00 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 173.9 (C), 172.4 (C), 164.0 (C), 108.6 (CH), 67.5 (CH), 64.0 (CH), 39.8 (CH), 33.4 $(CH₂)$, 31.5 (CH₂), 29.0 (CH₂), 26.9 (CH₂), 25.5 (CH₃), 22.4 (CH₂), 20.7 (CH₃), 17.8 (CH₃), 16.5 (CH₃), 13.9 (C), -4.4 (CH₃), -5.2 (CH₃). Anal. Calcd for C₂₁H₃₉NO₄Si: C, 63.43; H, 9.89; N, 3.52. Found: C, 63.69; H, 9.54; N, 3.74.

(S)-2-((R)-1-(tert-Butyldimethylsilyloxy)ethyl)-5-oxo-2,5-dihydro-1H-3-pyrrolyl (R)-2-Methyloctanoate (5m). The crude product was purified by column chromatography (silica gel, hexane/ EtOAc = 3/1–1/1) to give 5l (88.6 mg, 77%) as a colorless oil: $\left[\alpha\right]_{D}^{-24}$ +3.5 (c 0.769, CHCl3); IR (NaCl) 3226 (N−H), 3083 (N−H), 1777 (C=O), 1699 (C=O), 1624 (C=C) cm⁻¹; ¹H NMR (300 MHz, CDCl3) δ 6.54 (brs, 1H), 6.07 (s, 1H), 4.03−3.96 (m, 2H), 2.58 (m, 1H), 1.72 (m, 1H), 1.49 (m, 1H), 1.34−1.24 (m, 11H), 1.23 (d, J = 6.9 Hz, 3H), 0.89−0.82 (m, 12H), 0.04 (s, 3H), 0.01 (s, 3H); 13C NMR (75 MHz, CDCl₃) δ 173.8 (C), 172.3 (C), 164.1 (C), 108.6 (CH), 67.7 (CH), 64.0 (CH), 39.9 (CH), 33.2 (CH₂), 31.5 (CH₂), 29.0 (CH₂), 26.9 (CH₂), 25.5 (CH₃), 22.4 (CH₃), 20.7 (CH₃), 17.8 (CH₃), 16.6 (CH₃), 13.9 (C), -4.4 (CH₃), -5.2 (CH₃). Anal. Calcd for $C_{21}H_{39}NO_4Si$: C, 63.43; H, 9.89; N, 3.52. Found: C, 63.16; H, 9.65; N, 3.55.

(S)-2-((S)-sec-Butyl)-5-oxo-2,5-dihydro-1H-3-pyrrolyl Pivalate (5n). To a solution of tetramic acid 4b (80.0 mg, 0.515 mmol) in Et₂O (1.7 mL) were added Piv_2O (144 mg, 0.773 mmol) and pyridine (61.1 mg, 0.772 mmol). The solution was stirred at room temperature for 17 h until completion of the reaction (monitored by TLC). The reaction mixture was concentrated under reduced pressure, and the residue was purified by column chromatography (silica gel,

hexane/EtOAc = $1/1$) to give 5n (104 mg, 84%) as a yellow powder: dp 127−133 °C; [α]_D²⁷ +17.5 (c 1.32, CHCl₃); IR (NaCl) 3199 (N− H), 3077 (N−H), 2935(C−H), 1772 (C=O), 1682 (C=O), 1619 (C=C) cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 6.07 (s, 1H), 4.22 (brt, $J = 1.5$ Hz, 1H), 1.85 (m, 1H), 1.40−1.07 (m, 2H), 1.30 (s, 9H), 1.04 (d, J = 6.9 Hz, 3H), 0.87 (t, J = 7.5 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 174.2 (C), 174.0 (C), 165.6 (C), 108.1 (CH), 62.8 (CH), 39.5 (C), 36.2 (CH), 26.7 (CH₃), 23.0 (CH₂), 15.9 (CH₃), 11.8 (CH₃). Anal. Calcd for $C_{13}H_{21}NO_3$: C, 65.25; H, 8.84; N, 5.85. Found: C, 65.13; H, 8.66; N, 6.02.

General Procedure for the Preparation of the 3-Acyltetramic Acids. To a solution of 4-O-acyltetramic acid 5 (1.0 equiv) in dichloromethane were added $CaCl₂$ (1.5 equiv), DMAP (0.3 equiv), and Et₃N (1.2−1.5 equiv). After the suspension was stirred until completion of the reaction (monitored by TLC), the reaction was quenched by addition of 3% aqueous HCl, and the resulting mixture was extracted with EtOAc. The organic extracts were washed with water and brine, dried over anhydrous Na₂SO₄, filtered, and concentrated under reduced pressure. Purification of the crude product afforded 3-acyltetramic acid 6.

(S)-5-Benzyl-3- $(\alpha$ -hydroxybenzylidene)-4-oxo-2-pyrrolidone (6a). The crude product was washed with MeOH (5 mL) to give 6a (64.5 mg, 71%) as a white powder: dp 168−169 °C; IR (KBr) 3213 $(O-H)$, 3062 (N−H), 1702 (C=O), 1685 (C=O), 1669 (C=C) cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 8.20 (m, 2H), 7.64–7.48 (m, 3H), 7.35−7.21 (m, 5H), 6.86 (brs, 0.97H, Z-isomer), 6.21 (brs, 0.03H, E-isomer), 4.28 (dd, J = 2.7, 8.4 Hz, 0.03H, E-isomer), 4.05 (dd, $J = 3.6$, 10.1 Hz, 0.97H, Z-isomer), 3.32 (dd, $J = 3.6$, 13.8 Hz, 1H), 2.71 (dd, J = 10.1, 13.8 Hz, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 192.2 (C), 181.4 (C), 177.1 (C), 136.5 (C), 133.9 (C), 131.7 (CH), 129.7 (CH), 129.2 (CH), 128.9 (CH), 128.2 (CH), 127.2 (CH), 99.5 (C), 62.9 (CH), 38.3 (CH₂). Anal. Calcd for C₁₈H₁₅NO₃: C, 73.71; H, 5.15; N, 4.78. Found: C, 74.08; H, 5.09; N, 5.00.

(S)-5-((S)-sec-Butyl)-3-(α-hydroxybenzylidene)-4-oxo-2-pyrrolidone (6b). The crude product was purified by column chromatography (silica gel, hexane/EtOAc/AcOH = 83/16/1) to give 6b (78.9 mg, 77%, 6b/5-epimer = $83/17$) as an orange oil: IR (KBr) 3256 (O−H), 1685 (C=O), 1660 (C=O), 1596 (C=C) cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 8.21 (d, J = 7.2 Hz, 2H), 7.60 (m, 1H), 7.51 (m, 2H), 7.00 (brs, 1H), 4.16 (brs, 0.04H, 5-epimer, Eisomer), 4.06 (brd, $J = 3.6$ Hz, 0.14H, E-isomer), 3.92 (brd, $J = 2.7$ Hz, 0.13H, 5-epimer, Z-isomer), 3.84 (brd, J = 3.3 Hz, 0.69H, Z-isomer), 2.03 (m, 1H), 1.48−1.16 (m, 2H), 1.10−0.96 (m, 3H), 0.94−0.82 (m, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 193.4 (C), 180.5 (C), 177.8 (C), 133.8 (C), 131.8 (CH), 130.7 (CH), 129.7 (2CH), 128.2 (CH), 128.0 (CH), 100.6 (C), 66.4 (CH), 65.1 (CH), 37.1 (CH), 36.6 (CH), 26.8 (CH₂), 23.4 (2CH₂), 15.7 (CH₃), 15.4 (CH₃), 12.8 (CH₃), 11.6 $(CH₃)$. Anal. Calcd for $C₁₅H₁₇NO₃$: C, 69.48; H, 6.61; N, 5.40. Found: C, 69.19; H, 6.38; N, 5.44.

 $(S)-5-((S)-sec-Butyl)-3-(\alpha-hydroxy-4-methylbenzylidene)-4$ oxo-2-pyrrolidone (6c). The crude product was purified by column chromatography (silica gel, hexane/EtOAc/AcOH = 66/33/1) to give 6c (70.0 mg, 78%, 6c/5-epimer = 70/30) as a yellow oil: IR (KBr) 3235 (O–H), 1699 (C=O), 1657 (C=O), 1613 (C=C) cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 8.23−8.13 (m, 2H), 7.35−7.29 (m, 2H), 6.83 (s, 1H), 4.13 (brd, $J = 3.0$ Hz, 0.01H, 5-epimer, E-isomer), 4.04 $(brd, J = 4.2 \text{ Hz}, 0.11 \text{H}, E\text{-isomer}),$ 3.90 $(brd, J = 3.3 \text{ Hz}, 0.29 \text{H}, 5$ epimer, Z-isomer), 3.82 (brd, J = 3.3 Hz, 0.59H, Z-isomer), 2.43 (s, 3H), 2.11−1.96 (m, 1H) 1.51−1.19 (m, 2H), 1.06−0.80 (m, 6H); 13C NMR (75 MHz, CDCl₃) δ 193.6 (C), 181.8 (C), 181.7 (C), 176.7 (C), 176.4 (C), 137.2 (C), 137.1 (C), 130.8 (2C), 129.3 (2CH), 125.4 (2CH), 125.1 (2CH), 102.1 (C), 100.6 (C), 66.6 (CH), 65.4 (CH), 37.0 (CH), 36.5 (CH), 26.7 (CH₂), 23,6 (CH₂), 19.6 (CH₃), 15.5 (CH₃), 12.9 (CH₃), 11.6 (2CH₃). Anal. Calcd for C₁₆H₁₉NO₃: C, 70.31; H, 7.01; N, 5.12. Found: C, 69.92; H, 6.79; N, 5.21.

 $(S)-5-((S)-sec-Butyl)-3-(\alpha-hydroxy-4-chlorobenzylidene)-4$ oxo-2-pyrrolidone (6d). The crude product was purified by column chromatography (silica gel, hexane/EtOAc/AcOH = $66/33/1$) to give 6d (83.1 mg, 73%, 6d/5-epimer = 87/13) as a yellow powder: dp 91− 101 °C; IR (KBr) 3222 (O−H), 3075 (N−H), 1707 (C=O), 1661

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(C=O), 1590 (C=C) cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 8.30– 8.20 (m, 2H), 7.58 (brs, 1H), 7.47 (d, $J = 8.7$ Hz, 2H), 4.18 (brd, $J =$ 2.7 Hz, 0.03H, 5-epimer, E-isomer), 4.09 (brd, J = 3.6 Hz, 0.16H, Eisomer), 3.93 (brd, J = 2.7 Hz, 0.11H, 5-epimer, Z-isomer), 3.85 (brd, J = 2.7 Hz, 0.70H, Z-isomer), 2.04 (m, 1H), 1.10−0.96 (m, 3H), 0.94−0.81 (m, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 193.5 (2C), 179.0 (2C), 177.7 (C), 140.1 (CH), 132.1 (C), 131.4 (CH), 131.1 (CH), 130.2 (C), 128.8 (CH), 128.5 (CH), 128.4 (CH), 100.7 (C), 66.5 (CH) , 65.2 (CH), 63.7 (CH), 37.1 (CH), 37.0 (CH), 26.7 (CH₂), 23.8 (CH_2) , 23.4 (CH₂), 15.7 (CH₃), 15.3 (CH₃), 12.8 (CH₃), 11.6 (CH₃), 11.5 (CH₃). Anal. Calcd for C₁₅H₁₆ClNO₃: C, 61.33; H, 5.49; N, 4.77. Found: C, 61.13; H, 5.45; N, 4.59.

(S)-5-((S)-sec-Butyl)-3-(1-hydroxy-2-methylpropylidene)-4 oxo-2-pyrrolidone (6e). The crude product was purified by column chromatography (silica gel, hexane/EtOAc/AcOH = $90/9/1$) to give 6e (34.4 mg, 68%) as an orange powder: dp 83–89 °C; $[\alpha]_D^2$ ⁹ –58.1 $(c$ 1.03, CHCl₃); IR (KBr) 3207 (O–H), 3064 (N–H), 1715 (C= O), 1655 (C=O), 1610 (C=C) cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 7.12 (brs, 0.82H, Z-isomer), 6.70 (brs, 0.18H, E-isomer), 3.96 (brd, $J = 3.9$ Hz, 0.18H, E-isomer), 3.78 (brd, $J = 3.6$ Hz, 0.82H, Z-isomer), $3.74-3.62$ (m, 1H), 1.95 (m, 1H), 1.45−1.19 (m, 8H), 1.02 (d, J = 6.9 Hz, 3H), 0.91 (t, J = 7.2 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 197.7 (C), 195.0 (C), 193.2 (C), 176.3 (C), 100.3 (C), 66.8 (CH), 63.3 (CH), 37.0 (CH), 36.9 (CH), 31.8 (CH), 31.0 (CH), 23.9 (CH₂), 23.5 (CH₂), 18.4 (2CH₃), 15.3 (CH₃), 11.6 (CH₃), 11.5 (CH₃). Anal. Calcd for $C_{12}H_{19}NO_3$: C, 63.98; H, 8.50; N, 6.22. Found: C, 63.65; H, $8.11: N, 6.62.$

(S)-5-((S)-sec-Butyl)-3-(1-hydroxy-2-methylbutylidene)-4 oxo-2-pyrrolidone (6f). The crude product was purified by column chromatography (silica gel, hexane/EtOAc = $1/1$) to give 6f (35.5 mg, 74%) as a pale yellow oil: IR (KBr) 3236 (O–H), 1690 (C=O), 1658 (C=O), 1609 (C=C) cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 7.07 (brs, 0.8H, Z-isomer), 6.73 (brs, 0.2H, E-isomer), 3.95 (d, J = 3.6 Hz, 0.2H, E-isomer), 3.78 (d, J = 2.7 Hz, 0.8H, Z-isomer), 3.57 (m, 1H), 1.97 (m, 1H), 1.74 (m, 1H), 1.54 (m, 1H), 1.38 (m, 1H), 1.26 (m, 2H), 1.21−1.17 (m, 3H), 1.02 (d, J = 6.9 Hz, 3H), 0.95−0.87 (m, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 195.4 (2C), 192.7 (C), 192.5 (C), 176.4 (C), 176.3 (C), 101.6 (C), 101.6 (C), 101.4 (C), 67.0 (CH), 63.5 (CH), 37.6 (CH), 37.4 (CH), 26.9 (CH₂), 26.6 (CH₂), 23.6 (CH_2) , 23.5 (CH_2) , 16.6 (CH_3) , 16.4 (CH_3) , 15.7 $(2CH_3)$, 11.6 (2CH₃), 11.5 (2CH₃). Anal. Calcd for C₁₃H₂₁NO₃: C, 65.25; H, 8.84; N, 5.85. Found: C, 65.64; H, 9.00; N, 6.09.

(S)-5-((S)-sec-Butyl)-3-(cyclohexyl(hydroxy)methylene)-4 oxo-2-pyrrolidone (6g). The crude product was purified by column chromatography (silica gel, hexane/EtOAc/AcOH = $83/16/1$) to give 6g (89.6 mg, 72%) as an orange oil: $[\alpha]_D^{30}$ –54.5 (c 0.966, CHCl₃); IR (KBr) 3235 (O−H), 1706 (C=O), 1657 (C=O), 1608 (C=C) cm^{−1}; ¹H NMR (300 MHz, CDCl₃) δ 7.67 (brs, 0.8H, Z-isomer), 7.31 (brs, 0.2H, E-isomer), 3.95 (d, J = 3.3 Hz, 0.2H, E-isomer), 3.78 (d, J = 2.7 Hz, 0.8H, Z-isomer), 3.46−3.37 (m, 1H), 1.98 (m, 1H), 1.79−1.21 $(m, 12H)$, 1.00 $(d, J = 7.2 \text{ Hz}, 3H)$, 0.90 $(t, J = 7.2 \text{ Hz}, 0.6H, E$ isomer), 0.89 (t, J = 7.5 Hz, 2.4H, Z-isomer); ¹³C NMR (75 MHz, CDCl3) δ 195.0 (C), 192.5 (C), 176.4 (C), 100.4 (C), 67.0 (CH), 41.4 (CH) , 40.8 (CH), 37.0 (CH₂), 36.9 (CH₂), 28.6 (4CH₂), 25.4 (2CH₂), 25.3 (2CH₂) 23.9 (CH₂), 23.5 (CH₂), 15.6 (CH₃), 15.3 (CH₃), 11.6 (CH₃), 11.5 (CH₃). Anal. Calcd for C₁₅H₂₃NO₃: C, 67.90; H, 8.74; N, 5.28. Found: C, 67.62; H, 8.34; N, 5.52.

(S)-5-Benzyl-3-(1-hydroxy-2-methylbutylidene)-4-oxo-2-pyrrolidone (6h). The crude product was purified by column chromatography (silica gel, toluene/EtOAc = $3/1$) to give 6h (54.3) mg, 79%) as a yellow oil: IR (KBr) 3234 (O−H), 1706 (C=O), 1659 $(C=O)$, 1608 $(C=C)$ cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 7.33− 7.19 (m, 5H), 5.72 (brs, 0.9H, Z-isomer), 5.47 (brs, 0.1H, E-isomer), 4.18 (dd, $J = 3.6$, 9.0 Hz, 0.1H, E-isomer), 4.02 (dd, $J = 3.5$, 10.1 Hz, 0.9H, Z-isomer), 3.55 (m, 2H), 3.26 (m, 1H), 2.69 (m, 1H), 1.68 (m, 1H), 1.53 (m, 1H), 1.21−1.16 (m, 3H), 0.96−0.86 (m, 3H); 13C NMR (75 MHz, CDCl₃) δ 194.2 (C), 193.5 (C), 175.6 (C), 136.5 (C), 136.3 (C), 129.3 (CH), 129.2 (CH), 128.8 (2CH), 127.1 (CH), 100.5 (C), 100.4 (C), 63.4 (CH), 63.3 (CH), 38.1 (CH), 38.0 (CH), 37.6 $(2CH_2)$, 33.8 (CH_2) , 29.6 (CH_2) , 26.7 (CH_2) , 26.5 (CH_2) , 25.5

 $(CH₂)$, 24.8 (CH₂), 23.8 (CH₂), 23.4 (CH₂), 16.5 (CH₃), 16.4 (CH₃), 11.5 (2CH₃). Anal. Calcd for C₁₆H₁₉NO₃: C, 70.31; H, 7.01; N, 5.12. Found: C, 70.24; H, 7.23; N, 5.26.

(S)-3-(1-Hydroxy-2-methylbutylidene)-5-methyl-4-oxo-2 pyrrolidone (6i). The crude product was purified by column chromatography (silica gel, CHCl3/MeOH = 80/1−50/1−30/1) to give 6i (47.1 mg, 67%) as a yellow oil: IR (KBr) 3259 (O−H), 1685 $(C=0)$, 1656 $(C=0)$, 1618 $(C=C)$ cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 6.19 (brs, 0.8H, Z-isomer), 6.09 (brs, 0.2H, E-isomer), 4.06 $(q, J = 7.2$ Hz, 0.2H, E-isomer), 3.89 $(dq, J = 2.1, 6.9$ Hz, 0.8H, Zisomer), 3.64 (m, 0.2H, E-isomer), 3.55 (m, 0.8H, Z-isomer), 1.73 (m, 1H), 1.55 (m, 1H), 1.41 (d, J = 6.9 Hz, 0.6H, E-isomer), 1.37 (d, J = 6.9 Hz, 2.4H, Z-isomer), 1.22−1.18 (m, 3H), 0.97−0.90 (m, 3H); 13C NMR (75 MHz, CDCl₃) δ 195.6 (C), 193.6 (C), 175.6 (C), 99.8 (C), 57.7 (CH), 54.4 (CH), 38.0 (CH), 37.6 (CH), 29.6 (CH₂), 26.7 (CH_2) , 17.4 (CH_3) , 16.5 (CH_3) , 11.5 $(2CH_3)$. Anal. Calcd for $C_{10}H_{15}NO_3$: C, 60.90; H, 7.67; N, 7.10. Found: C, 60.57; H, 7.27; N, 7.20.

(S)-5-((tert-Butyldimethylsilyloxy)methyl)-3-(1-hydroxy-2 methylbutylidene)-4-oxo-2-pyrrolidone (6j). The crude product was purified by column chromatography (silica gel, hexane/EtOAc = 1/1) to give 6j (28.0 mg, 72%) as an orange oil: IR (KBr) 3438 (N− H), 3234 (O−H), 1708 (C=O), 1661 (C=O), 1607 (C=C) cm⁻¹;
¹H NMR (300 MHz, CDCL) δ 6.43 (brs. 0.9H, Z-isomer), 6.18 (brs. ¹H NMR (300 MHz, CDCl₃) δ 6.43 (brs, 0.9H, Z-isomer), 6.18 (brs, 0.1H, E-isomer), 4.10−3.45 (m, 4H), 1.71 (m, 1H), 1.53 (m, 1H), 1.21−1.16 (m, 3H), 0.95−0.88 (m, 3H), 0.86 (s, 9H), 0.05 (s, 6H); ¹³C NMR (75 MHz, CDCl₃) δ 193.1 (C), 192.0 (C), 176.2 (C), 100.1 (C) , 64.1 (CH), 63.2 (CH₂), 61.2 (CH₂), 37.7 (CH), 26.6 (C), 25.6 (CH₃), 18.0 (CH₂), 16.5 (CH₃), 11.5 (CH₃), -5.7 (2CH₃). Anal. Calcd for $C_{16}H_{29}NO_4Si$: C, 58.68; H, 8.93; N, 4.28. Found: C, 58.50; H, 8.56; N, 4.23.

(S)-5-((R)-1-(tert-Butyldimethylsilyloxy)ethyl)-3-(1-hydroxy-2-methylbutylidene)-4-oxo-2-pyrrolidone (6k). The crude product was purified by column chromatography (silica gel, hexane/EtOAc = 1/2) to 6k (63.4 mg, 86%) as a yellow oil: IR (KBr) 3449 (N−H), 3239 (O−H), 1709 (C=O), 1660 (C=O), 1608 (C=C) cm⁻¹; ¹H NMR (300 MHz, CDCl3) δ 7.17 (brs, 0.8H, Z-isomer), 7.13 (brs, 0.2H, E-isomer), 4.17 (m, 2H), 3.79 (m, 0.2H, E-isomer), 3.62 (m, 0.8H, Z-isomer), 3.48 (m, 2H), 1.67 (m, 2H), 1.48 (m, 2H), 1.30− 1.10 (m, 12H), 0.92−0.77 (m, 24H), 0.05--0.05 (m, 6H); 13C NMR (75 MHz, CDCl₃) δ 193.8 (C), 193.5 (C), 192.2 (C), 176.6 (C), 101.5 (C), 101.1 (C), 68.0 (CH), 67.6 (CH), 65.2 (CH), 37.7 (CH), 37.3 (CH), 27.0 (C), 25.9 (CH₂), 25.5 (2CH₃), 21.1 (CH₃), 21.0 (CH₃), 17.7 (CH₃), 16.8 (CH₃), 16.0 (CH₃), 11.7 (CH₃), 11.4 (CH₃), −4.4 $(2CH_3)$, -5.3 (CH₃), -5.4 (CH₃). Anal. Calcd for C₁₇H₃₁NO₄Si: C, 59.79; H, 9.15; N, 4.10. Found: C, 59.72; H, 9.12; N, 4.49.

(S)-5-((R)-1-(tert-Butyldimethylsilyloxy)ethyl)-3-((S)-1-hydroxy-2-methyloctylidene)-4-oxo-2-pyrrolidone (6l). The crude product was purified by column chromatography (silica gel, hexane/ EtOAc = 3/1) to give 6l (37.7 mg, 84%) as a colorless oil: $[\alpha]_D^{\text{17}}$ −64.1 (c 0.636, CHCl3); IR (NaCl) 3235 (N−H), 3068 (N−H), 1694 (C=O), 1660 (C=O), 1615 (C=C) cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 5.94 (brs, 0.8H, Z-isomer), 5.67 (brs, 0.2H, E-isomer), 4.14 $(m, 1H)$, 3.78 (dd, J = 0.6, 3.9 Hz, 0.2H, E-isomer), 3.70 $(m, 1H)$, 3.61 (dd, J = 0.6, 4.2 Hz, 0.8H, Z-isomer), 3.59 (m, 1H), 1.66 (m, 1H), 1.47 $(m, 1H)$, 1.32−1.22 $(m, 8H)$, 1.28 $(d, J = 6.3$ Hz, 3H), 1.19 $(d, J = 6.9)$ Hz, 0.6H, E-isomer), 1.16 (d, J = 6.6 Hz, 2.4H, Z-isomer), 0.86 (t, J = 7.0 Hz, 3H), 0.82 (s, 9H), 0.05 (s, 3H), 0.01 (s, 3H); 13C NMR (75 MHz, CDCl₃) δ 193.6 (C), 192.5 (C), 176.6 (C), 104.4 (C), 101.3 (C), 68.0 (CH), 67.6 (CH), 67.4 (CH), 65.2 (CH), 36.1 (CH), 35.9 (CH), 34.0 (CH₂), 33.5 (CH₂), 31.6 (CH₂), 29.1 (CH₂), 27.1 (CH₂), 25.5 (CH₃), 22.5 (CH₂), 21.1 (CH₃), 20.9 (CH₃), 17.7 (C), 16.9 (CH₃), 16.5 (CH₃), 13.9 (CH₃), -4.4 (CH₃), -5.4 (CH₃). Anal. Calcd for C₂₁H₃₉NO₄Si: C, 63.43; H, 9.89; N, 3.52. Found: C, 63.08; H, 9.55; N, 3.69.

(S)-5-((R)-1-(tert-Butyldimethylsilyloxy)ethyl)-3-((R)-1-hydroxy-2-methyloctylidene)-4-oxo-2-pyrrolidone (6m). The crude product was purified by column chromatography (silica gel, hexane/EtOAc = $3/1$) to give 6m (35.7 mg, 79%) as a colorless oil: $[\alpha]_{D}^{23}$ –86.8 (c 1.13, CHCl₃); IR (NaCl) 3236 (N–H), 3068 (N–H),

1694 (C=O), 1660 (C=O), 1614 (C=C) cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 6.00 (brs, 0.75H, Z-isomer), 5.73 (brs, 0.25H, E-isomer), 4.14 (m, 0.25H, E-isomer), 4.12 (m, 0.75H, Z-isomer), 3.77 $(dd, J = 0.6, 3.9 Hz, 0.25H, E-isomer), 3.70 (m, 0.25H, E-isomer), 3.60$ $(d, J = 4.5 \text{ Hz}, 0.75 \text{ H}, Z\text{-isomer})$, 3.54 (m, 0.75H, Z-isomer), 1.68 (m, 1H), 1.45 (m, 1H), 1.32−1.24 (m, 8H), 1.29 (d, J = 6.6 Hz, 3H), 1.18 (d, $J = 6.9$ Hz, 2.25H, Z-isomer), 1.17 (d, $J = 6.9$ Hz, 0.75H, Eisomer), 0.87 (t, $J = 7.2$ Hz, $CH₃$), 0.82 (s, 6.75H, Z-isomer), 0.80 (s, 2.25H, E-isomer), 0.05 (s, 2.25H, Z-isomer), 0.04 (s, 0.75H, Eisomer), 0.01 (s, 2.25H, Z-isomer), −0.02 (s, 0.75H, E-isomer); 13C NMR (75 MHz, CDCl₃) δ 201.4 (C), 195.2 (C), 193.3 (C), 192.8 (C), 176.5 (C), 169.4 (C), 104.6 (C), 100.9 (C), 100.6 (C), 68.0 (CH), 67.5 (CH), 65.2 (CH), 39.4 (CH), 36.3 (CH), 35.9 (CH₂), 33.6 (CH_2) , 32.9 (CH₂), 31.6 (CH₂), 29.3 (CH₂), 29.2 (CH₂), 27.3 (CH₂), 27.2 (CH₂), 25.5 (CH₃), 22.5 (CH₂), 21.1 (CH₃), 20.8 (CH₃), 17.7 (C), 17.2 (CH₃), 17.0 (CH₃), 13.9 (CH₃), -4.4 (2CH₃), -5.2 (CH₃), -5.5 (CH₃). Anal. Calcd for C₂₁H₃₉NO₄Si: C, 63.43; H, 9.89; N, 3.52. Found: C, 63.35; H, 9.89; N, 3.52.

(3R)-2-Benzyloxycarbonylamino-3-(tert-butyldimethylsilyloxy) butanoic Acid (8). To a solution of a 56:44 mixture of L-threonine and D-allothreonine (1.00 g, 12.6 mmol) in saturated aqueous NaHCO₃ solution (35 mL) was added carbobenzoxy chloride (CbzCl) (2.15 g, 12.6 mmol) at room temperature. After stirring for 18 h at the same temperature, the reaction mixture was poured into water. The aqueous layer was washed with $Et₂O$, acidified with concentrated HCl, and extracted with EtOAc. The combined organic extracts were dried over anhydrous $MgSO₄$ and concentrated under reduced pressure. To a solution of the residue (1.73 g) in DMF (27 mL) were added imidazole (1.16 g, 17.1 mmol), DMAP (100 mg, 0.819 mmol), and TBSCl (2.27 g, 15.0 mmol) at room temperature. After stirring for 18 h at the same temperature, the reaction mixture was poured into saturated aqueous NH4Cl solution and extracted with EtOAc. The combined organic extracts were washed with 3% aqueous HCl and brine, dried over anhydrous $Na₂SO₄$, and concentrated under reduced pressure. The residue was roughly purified by column chromatography (silica gel, hexane/EtOAc = $1/1$) to afford the crude sample of 8. To a solution of the crude product in a 1:1 mixture of water and methanol (6.0 mL) was added an aqueous solution of K_2CO_3 (900 mg, in 2.0 mL). After stirring for 3 h, the solution was acidified to pH 3 with 3 M aqueous HCl. The reaction mixture was poured into EtOAc and extracted. The organic layer was dried over MgSO4, filtered, and concentrated under reduced pressure. The crude product was purified by column chromatography (silica gel, hexane/ EtOAc = $1/1$) to afford a diastereomeric mixture of 8 (1.80 g, 58%, 2S/2R = 56/44): IR (NaCl) 3400 (N−H), 3091 (O−H), 3067 (N− H), 1755 (C=O), 1691 (C=O) cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 7.37–7.29 (m, 10H), 5.50 (brd, J = 8.7 Hz, 1H), 5.47 (brd, J = 9.3 Hz, 1H), 5.12−5.10 (m, 4H), 4.46 (m, 1H), 4.33−4.30 (m, 2H), 4.13 (m, 1H), 1.28 (d, $J = 6.3$ Hz, 3H), 1.20 (d, $J = 6.3$ Hz, 3H), 0.86 (s, 9H), 0.85 (s, 9H), 0.09 (s, 3H), 0.07 (s, 3H), 0.05 (s, 3H), 0.04 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 174.5 (C), 156.7 (C), 136.1 (C), 128.7 (CH), 128.6 (CH), 128.4 (CH), 128.3 (CH), 128.2 (2CH), 69.7 (CH), 68.5 (CH), 67.3 (CH₂), 67.1 (CH₂), 59.8 (CH) , 59.3 (CH), 25.6 (CH₃), 25.5 (CH₃), 20.2 (CH₃), 19.6 (CH₃), 17.8 (C), 17.7 (C), -4.7 (2CH₃), -5.3 (CH₃), -5.4 (CH₃). Anal. Calcd for $C_{18}H_{29}NO_5Si$: C, 58.83; H, 7.95; N, 3.81. Found: C, 58.87; H, 7.67; N, 3.58.

5-((R)-1-(tert-Butyldimethylsilyloxy)ethyl)-4-oxo-2-pyrrolidone (9). To a solution of Meldrum's acid (777 mg, 5.39 mmol) and 8 (1.80 g, 4.90 mmol) in dichloromethane (25 mL) at 0 $^{\circ}$ C were added EDCI (1.13 g, 5.88 mmol) and DMAP (838 mg, 6.86 mmol). After stirring for 16 h at room temperature, the reaction mixture was poured into EtOAc (70 mL) and extracted. The organic layer was washed with 5% aqueous NaHSO₄ solution and brine, dried over $Na₂SO₄$, and filtered. The filtrate was refluxed for 2 h until completion of the reaction (monitored by TLC) and concentrated under reduced pressure. The residue was roughly purified by column chromatography (silica gel, CHCl₃/MeOH = $10/1$) to give the crude product (1.90 g) as a yellow oil. A suspension containing the crude product and Pd/C (100 mg) in EtOAc (10 mL) was stirred under hydrogen atmosphere for 4 h. The reaction mixture was filtered through a pad of Celite, and the residue was washed with EtOAc. The filtrate was concentrated under reduced pressure, and the residue was purified by column chromatography (silica gel, hexane/EtOAc = $3/1-2/1-1/1-1/2$) to give 9 (636 mg, 50%, 3 steps, $5R/5S = 61/39$) as a white powder: IR (NaCl) 3201 (N−H), 3108 (N−H), 1768 (C=O), 1702 (C=O) cm[−]¹ ; 1 H NMR (300 MHz, CDCl3) δ 7.85 (brs, 1H, 5S-isomer), 7.41 (brs, 1H, 5R-isomer), 4.21 (dq, J = 2.7, 6.6 Hz, 1H, 5S-isomer), 4.06 (dq, J = 2.4, 6.3 Hz, 1H, 5R-isomer), 3.90 (brs, 1H, 5R-isomer), 3.80 (brd, J = 2.6 Hz, 1H, 5S-isomer), 2.91 (s, 2H, 5S-isomer), 2.89 (brs, 2H, 5R-isomer), 1.24 (d, J = 6.6 Hz, 3H, 5S-isomer), 1.23 (d, J = 6.3 Hz, 3H, 5R-isomer), 0.84 (s, 9H, 5R-isomer), 0.81 (s, 9H, 5S-isomer), 0.06 (s, 3H, 5R-isomer), 0.05 (s, 3H, 5R-isomer), 0.04 (s, 3H, 5Sisomer), -0.03 (s, 3H, 5S-isomer); ¹³C NMR (75 MHz, CDCl₃) δ 206.8 (C), 205.6 (C), 172.8 (C), 172.3 (C), 70.2 (CH), 69.6 (CH), 69.2 (CH), 68.1 (CH), 41.8 (CH₂), 41.3 (CH₂), 25.6 (CH₃), 25.5 $(CH₃)$, 20.3 (CH₃), 18.5 (CH₃), 17.7 (2C), −4.5 (CH₃), −5.0 (CH₃), −5.1 (CH₃), −5.4 (CH₃). Anal. Calcd for C₁₂H₂₃NO₃Si: C, 55.99; H, 9.01; N, 5.44. Found: C, 56.34; H, 8.80; N, 5.81.

(R)-2-((R)-1-(tert-Butyldimethylsilyloxy)ethyl)-5-oxo-2,5-dihydro-1H-3-pyrrolyl (S)-2-Methyloctanoate (10). The crude product was purified by column chromatography (silica gel, hexane/ toluene/EtOAc = $1/1/1$, $CH_2Cl_2/CHCl_3/MeOH = 10/10/1$) to give **10** (127 mg, 27%) as a colorless oil: $[\alpha]_D^2$ ⁴ +29.7 (c 1.37, CHCl₃); IR (NaCl) 3207 (N−H), 3091 (N−H), 1778 (C=O), 1701 (C=O), 1621 (C=C) cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 6.05 (brs, 1H), 5.69 (brs, 1H), 4.32 (d, $J = 3.3$ Hz, 1H), 4.16 (m, 1H), 2.63 (m, 1H), 1.70 (m, 1H), 1.53 (m, 1H), 1.34−1.24 (m, 8H), 1.24 (d, J = 6.9 Hz, 3H), 0.99 (d, J = 6.3 Hz, 3H), 0.91−0.88 (m, 12H), 0.10 (s, 3H), 0.09 $(s, 3H)$; ¹³C NMR (75 MHz, CDCl₃) δ 173.1 (C), 172.1 (C), 163.1 (C) , 108.6 (CH) , 67.6 (CH) , 63.2 (CH) , 39.8 (CH) , 33.6 $(CH₂)$, 31.6 $(CH₂)$, 29.1 (CH₂), 27.0 (CH₂), 25.7 (CH₃), 22.6 (CH₃), 17.9 (CH₃), 16.6 (CH₃), 16.2 (CH₃), 14.0 (C), -4.7 (CH₃), -5.0 (CH₃). Anal. Calcd for C₂₁H₃₉NO₄Si: C, 63.43; H, 9.89; N, 3.52. Found: C, 63.63; H, 9.52; N, 3.83.

(R)-5-((R)-1-(tert-Butyldimethylsilyloxy)ethyl)-3-((S)-1-hydroxy-2-methyloctylidene)-4-oxo-2-pyrrolidone (11). The crude product was purified by column chromatography (silica gel, hexane/ EtOAc = $3/1-1/1$) to give 11 (40.1 mg, 93%) as a colorless oil: $[\alpha]_{\text{D}}^{20}$ +68.8 (c 0.904, CHCl₃); IR (NaCl) 3229 (N–H), 3069 (N– H), 1716 (C=O), 1660 (C=O), 1610 (C=C) cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 6.27 (brs, 0.85H, Z-isomer), 6.06 (brs, 0.15H, Eisomer), 4.26 (m, 1H), 4.07 (d, J = 3.3 Hz, 0.15H, E-isomer), 3.94 (d, J = 3.3 Hz, 0.85H, Z-isomer), 3.71 (m, 0.15H, E-isomer), 3.60 (m, 0.85H, Z-isomer), 1.67 (m, 1H), 1.46 (m, 1H), 1.30−1.21 (m, 8H), 1.19 (d, J = 6.9 Hz, 0.45H, E-isomer), 1.17 (d, J = 6.9 Hz, 2.55H, Zisomer), 1.10 (d, $J = 6.3$ Hz, 0.45H, E-isomer), 1.03 (d, $J = 6.0$ Hz, 2.55H, Z-isomer), 0.88–0.86 (m, 12H), 0.09 (s, 3H), 0.08 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 193.4 (C), 192.9 (C), 176.4 (C), 101.4 (C) , 68.3 (CH), 67.5 (CH), 36.1 (CH), 33.8 (CH₂), 33.6 (CH₂), 31.6 (CH_2) , 29.2 (CH₂), 29.1 (CH₂), 27.1 (CH₂), 25.6 (CH₃), 22.5 (CH₂), 17.8 (C), 17.4 (CH₃), 16.9 (CH₃), 16.7 (CH₃), 14.0 (CH₃), -4.7 (CH_3) , -5.0 (CH₃). Anal. Calcd for C₂₁H₃₉NO₄Si: C, 63.43; H, 9.89; N, 3.52. Found: C, 63.23; H, 9.50; N, 3.84.

(R)-5-((R)-1-(tert-Butyldimethylsilyloxy)ethyl)-3-((S)-1-hydroxy-2-methyloctylidene)-1-methyl-4-oxo-2-pyrrolidone (12). To a solution of 11 (57.5 mg, 0.145 mmol) in THF (2.0 mL) was added NaHMDS (1.09 M solution in THF, 0.29 mL, 0.32 mmol) at −78 °C. After stirring for 20 min at −78 °C, MeI (0.090 mL, 1.5 mmol) was added to the mixture. The resulting mixture was warmed to −40 °C and stirred for 10 h. It was quenched with 5% aqueous tartaric acid solution (5 mL) and extracted with EtOAc. Combined organic extracts were washed with H_2O and brine, dried over Na_2SO_4 , filtered, and concentrated under reduced pressure. The crude product was purified by column chromatography (silica gel, hexane/EtOAc = 10/1−7/1−5/1−1/1) to give 12 (37.0 mg, 62%) as a colorless oil: $[\alpha]_D^{20}$ +32.9 (c 0.371, CHCl₃); IR (NaCl) 1712 (C=O), 1653 (C= O), 1619 (C=C) cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 4.32 (dq, J = 2.4, 6.3 Hz, 1H), 3.80 (d, J = 2.1 Hz, 0.1H, E-isomer), 3.71 (d, J = 2.4 Hz, 0.9H, Z-isomer), 3.57 (m, 1H), 3.08 (s, 2.7H, Z-isomer), 3.03 (s, 0.3H, E-isomer), 1.65 (m, 1H), 1.44 (m, 1H), 1.29−1.20 (m, 8H), 1.16 (d, J = 6.6 Hz, 3H), 1.05 (d, J = 6.3 Hz, 3H), 0.88–0.85 (m, 12H), 0.097 (s, 3H), 0.088 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 192.1 (C), 191.7 (C), 174.4 (C), 101.1 (C), 72.4 (CH), 69.0 (CH), 36.0 (CH), 33.7 (CH₂), 31.6 (CH₂), 29.1 (CH₂), 28.3 (CH₃), 27.1 (CH_2) , 25.6 (CH₃), 22.5 (CH₂), 17.8 (C), 17.4 (CH₃), 17.0 (CH₃), 13.9 (CH₃), -4.8 (CH₃), -5.2 (CH₃). Anal. Calcd for C₂₂H₄₁NO₄Si: C, 64.19; H, 10.04; N, 3.40. Found: C, 64.57; H, 9.66; N, 3.79.

(+)-Penicillenol A_2 (2). A solution of 12 (37.0 mg, 0.0899 mmol) in 2% HCl−MeOH (3.0 mL) was stirred at 0 °C for 1 h. The mixture was warmed to room temperature and stirred for an additional 1.5 h. After removal of the solvent under vacuum, the residue was purified by column chromatography (silica gel, $CHCl₃/MeOH = 40/1$). A solution of isolated material in EtOAc (10 mL) was washed with 5% aqueous tartaric acid (10 mL) and brine, dried over Na_2SO_4 , filtered, and concentrated under reduced pressure to give 2 (17.4 mg, 65%) as a colorless oil: $\left[\alpha \right]_{\text{D}}{}^{22}$ +49.2 (c 0.137, MeOH, averaged value for 10 measurements); $[\alpha]_D^{22}$ +27.8 (c 0.864, CHCl₃); IR (NaCl) 3413 (O−H), 2960 (C−H), 2928 (C−H), 2857 (C−H), 1711 (C O), 1651 (C=O), 1615 (C=C) cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 4.30 (dq, J = 3.3, 6.3 Hz, 1H), 3.84 (d, J = 3.0 Hz, 0.1H, E-isomer), 3.75 (d, J = 3.3 Hz, 0.9H, Z-isomer), 3.58 (m, 1H), 3.11 (s, 2.7H, Zisomer), 3.06 (s, 0.3H, E-isomer), 1.66 (m, 1H), 1.46 (m, 1H),1.33− 1.23 (m, 8H), 1.21 (d, $J = 6.6$ Hz, 3H), 1.17 (d, $J = 6.6$ Hz, 3H), 0.86 $(t, J = 6.6 \text{ Hz}, 3\text{H})$; ¹³C NMR (75 MHz, CDCl₃) δ 192.4 (C), 192.4 (C), 174.6 (C), 100.8 (C), 71.5 (CH), 68.2 (CH), 36.0 (CH), 33.7 $(CH₂)$, 31.6 $(CH₂)$, 29.1 $(CH₂)$, 28.4 $(CH₃)$, 27.0 $(CH₂)$, 22.4 $(CH₂)$, 17.5 (CH₃), 17.0 (CH₃), 13.9 (CH₃); Anal. Calcd for C₁₆H₂₇NO₄: C, 64.62; H, 9.15; N, 4.71. Found: C, 64.67; H, 8.83; N, 4.50.

■ ASSOCIATED CONTENT

S Supporting Information

 1 H and 13 C NMR spectra of all the new compounds. This material is available free of charge via the Internet at http:// pubs.acs.org.

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Notes

The authors declare no competing fi[nancial interest.](mailto:tchyoda@ipc.shizuoka.ac.jp)

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(14) The epimeric isomer of 5b was not generated when the reactions of 5b were performed either using Et_3N or $CaCl₂$.

(15) Despite the limited solubilities in CH_2Cl_2 , we cannot exclude a possibility that the metal salts may act as a Lewis acid to accelerate the attack of DMAP to the vinyl esters.

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(18) One may expect that (E)-isomer included in the product mixture may cause a decrease in the optical rotation value. However, this interpretation can be substantially ruled out, considering that the literature data should be obtained with a mixture of the geometrically isomeric penicillenols generated in situ by inevitable equilibration that we observed in the ${}^{1}\text{H}$ NMR spectrum of the CDCl₃ solution. For tautomeric equilibrium of 3-acyltetramic acids in solutions, see: (a) Yamaguchi, T.; Saito, K.; Tsujimoto, T.; Yuki, H. Bull. Chem. Soc.

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(19) In fact, we observed that the optical rotation values ranged irregularly between +41.0 and +55.2 for 10 measurements performed on this compound.

(20) The structural and stereochemical integrity of the synthetic penicillenol A_2 were unequivocally confirmed by ¹H and ¹³C NMR. For more details on these observations and those for penicillenol $A₁$, see the Supporting Information.